



UNITED STATES ENVIRONMENTAL PROTECTION AGENCY
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OFFICE OF
PREVENTION, PESTICIDES
AND TOXIC SUBSTANCES

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MEMORANDUM

Date: 10/11/01

SUBJECT: Tier 1 Drinking Water Estimated Environmental
Concentrations for Urea.

To: Joseph Nevola
Special Review Branch
Special Reregistration Division (7509C)
and
Rebecca Daiss
Reregistration Branch 4
Health and Effects Division (7509C)

FROM: Ibrahim Abdel-Saheb/Agronomist
Environmental Risk Branch II
Environmental Fate and Effects Division (7507C)

PEER REVIEW: Sid Abel/Environmental Scientist
Environmental Risk Branch II
Environmental Fate and Effects Division (7507C)

THRU : Tom Bailey, Branch Chief
Environmental Risk Branch II
Environmental Fate and Effects Division (7507C)

This memorandum transmits the Tier I estimated drinking water concentrations for urea use on citrus. The FQPA Index Reservoir Screening Tool (FIRST)¹ was used to estimate these concentrations. The GENERIC Estimated Environmental Concentration (GENEEC 2.0)² was also used to estimate the surface water concentrations for urea to establish risk to aquatic organisms when used as an inert ingredient.

The SCI-GROW³ model was used to estimate groundwater drinking water concentrations for urea. Modeling results are shown in Table 1.

Table 1. Estimated environmental concentrations (ppm) of urea in surface and groundwater.				
Scenario	peak	long term average	use(s) modeled	PCA
Surface water (FIRST)	4.85	0.01	6 application @ 80 lb/acre on citrus	0.87
Surface water (GENEEC)	3.58	0.04	6 application @ 80 lb/acre	
Groundwater	0.001		6 application @ 80 lb/acre	

Environmental Fate and Transport Assessment

EFED has no fate data for Urea. Available data from literature reviews shows that urea degrades rapidly in most soils⁴⁻⁶. In general, urea is rapidly hydrolyzed to ammonium through soil urease activity. In various soils, the hydrolysis may near completion within 24 hrs⁴; however, the rate of hydrolysis can be much slower depending upon soil type, moisture content, and urea formulation. For example, increasing the pellet size of urea fertilizers can decrease the urea decomposition rate from days to weeks. Soil adsorption studies have demonstrated that urea adsorbs very weakly to soil⁷; therefore, leaching is possible. Ultimate urea degradation produces ammonia and CO₂ as volatile products⁸.

Biodegradation is expected to be the major fate process in the aquatic ecosystem. Various screening studies have demonstrated that urea can biodegrade readily⁹⁻¹³ with the release of CO₂ and ammonia. The rate of biodegradation generally decreases with decreasing temperatures¹²; under cold winter-like conditions, biodegradation may be relatively slow (0-6% per day)¹². The presence of

naturally-occurring phytoplankton increases the degradation rate^{10,13} because phytoplankton use urea as a nitrogen source¹⁰ and because urea is decomposed by phytoplankton photosynthesis¹³; in phytoplankton-rich waters, degradation occurs much faster in sunlight than in the dark¹³.

Abiotic hydrolysis of urea occurs very slowly in relation to biotic hydrolysis¹⁴. Abiotic hydrolysis yields ammonium carbamate which decomposes to form CO₂ and ammonia¹⁴; the enzyme urease catalyzes urea hydrolysis.

In one photodegradation study using a silica gel adsorbent⁹ only 0.2% of applied urea photomineralized after a 17-hr irradiation with a UV lamp (>290 nm).

The adsorption of urea was measured in six different British soils with organic carbon contents ranging from 1.76 to 36.5%; no adsorption was measurable in five of the soils¹⁵; in the sixth soil (36.5% organic carbon), a K_{oc} of 8 can be determined from the measured Freundlich isotherm¹⁶.

Surface Water

Monitoring

At the present time, the EFED has no monitoring data on the concentrations of urea in surface water.

Modeling

Surface water concentration estimates were modeled for the use of urea on citrus using FIRST and GENEEC Tier I models. The input parameters used in simulations are shown in Tables 2 and 3.

Table 2. Input parameters for Tier I model FIRST used for Urea.

Parameter	calculations/value	source
Crop name	citrus	EPA Reg. Lable No. 688915.
application rate (lb ai/acre)	81	EPA Reg. Lable No. 688915.
interval between applic. (day)	10	EPA Reg. Lable No. 688915.
Max No. application	6	EPA Reg. Lable No. 688915.
PCA factor (decimal)	0.87 (default)	Effland et al ¹⁷ (2000).
Koc (mL/g)	8	Hance (1965).
soil aerobic met. $t_{1/2}$ (d)	1 X 3	Scheunert I. (1987); FIRST User Manual.
pesticide to be wetted-in ?	No	EPA Reg. Lable No. 688915
method of application	aerial	EPA Reg. Lable No. 688915.
solubility (mg/L)	5.45×10^5	Yalkowsky S.H. (1989) ¹⁸ .
aerobic aquatic met. $t_{1/2}$ (d)	0.042 (assumed to be 1 hour: readily degraded)	Freitag D. (1985).
hydrolysis (pH 7) $t_{1/2}$ (d)	1	Sankhayan et al. (1976).
aqueous photolysis $t_{1/2}$ (d)	stable (0.2% < degraded after 17 hours of radiation)	Freitag et al. (1985).

Table 3. Input parameters for GENEEC 2.0 modeling of urea.

Parameter	calculations/value	source
Crop name	citrus	EPA Reg. Lable No. 688915.
application rate (lb ai/acre)	81	EPA Reg. Lable No. 688915.
interval between applic. (day)	10	EPA Reg. Lable No. 688915.
Max No. application	6	EPA Reg. Lable No. 688915.
Koc (mL/g)	8	Hance (1965).
soil aerobic met. $t_{1/2}$ (d)	1 X 3	Scheunert I. (1987); FIRST User Manual.
pesticide to be wetted-in ?	No	EPA Reg. Lable No. 688915
method of application	aerial	EPA Reg. Lable No. 688915.
Aerial droplet size distribution	fine to medium (default)	GENEEC Users Manual.
solubility (mg/L)	5.45×10^5	Yalkowsky (1989).
aerobic aquatic met. $t_{1/2}$ (d)	0.042 (assumed to be 1 hour: readily degraded)	Freitag (1985).

hydrolysis (pH 7) $t_{1/2}$ (d)	1	Sankhayan and Shukla (1976) .
aqueous photolysis $t_{1/2}$ (d)	stable (0.2% < degraded after 17 hours of radiation)	Freitag (1985) .

Groundwater

Monitoring

EFED has no monitoring data on the concentrations of urea in groundwater.

Modeling

The SCI-GROW model was used to estimate potential groundwater concentrations. SCI-GROW is a screening model for ground water. It is based on a regression approach which relates the concentrations found in ground water in Prospective Ground Water studies to aerobic soil metabolism rate and soil-water partitioning properties of the chemical.

The input and output files used in SCI-GROW are shown in Appendix I.

References

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APPENDIX I

FIRST output file

RUN No. 1 FOR urea ON citrus * INPUT VALUES *

RATE (#/AC) ONE (MULT)	No. APPS & INTERVAL	SOIL Koc	SOLUBIL (PPM)	APPL TYPE (%DRIFT)	%CROPPED AREA	INCORP (IN)
81.000 (89.921)	6 10	8.0*****		AERIAL (16.0)	87.0	.0

FIELD AND RESERVOIR HALFLIFE VALUES (DAYS)

METABOLIC COMBINED FIELD)	DAYS UNTIL RAIN/RUNOFF	HYDROLYSIS (RESERVOIR)	PHOTOLYSIS (RES.-EFF)	METABOLIC (RESER.)	(RESER.)
3.00	2	N/A	30.00- 3720.00	.04	.04

UNTREATED WATER CONC (MILLIGRAMS/LITER (PPM)) Ver 1.0 AUG 1, 2001

PEAK DAY (ACUTE) CONCENTRATION	ANNUAL AVERAGE (CHRONIC) CONCENTRATION
4.848	.010

GENEEC 2.0 input and output files

RUN No.	1	FOR UREA	ON	CITRUS	* INPUT VALUES *			
RATE (#/AC)	No. APPS & INTERVAL	SOIL Koc	SOLUBIL (PPM)	APPL TYPE (%DRIFT)	NO-SPRAY (FT)	INCORP (IN)		
81.000 (108.000)	10	6	8.0	*****	AERL_B (13.0)	.0	.0	

FIELD AND STANDARD POND HALFLIFE VALUES (DAYS)

METABOLIC COMBINED (FIELD) (POND)	DAYS UNTIL RAIN/RUNOFF	HYDROLYSIS (POND)	PHOTOLYSIS (POND-EFF)	METABOLIC (POND)
3.00	2	N/A	30.00- 3720.00	.04

GENERIC EECs (IN MILLIGRAMS/LITER (PPM)) Version 2.0 Aug 1, 2001

PEAK GEEC	MAX 4 DAY AVG GEEC	MAX 21 DAY AVG GEEC	MAX 60 DAY AVG GEEC	MAX 90 DAY AVG GEEC
3.58	.90	.17	.06	.04

SCI-GROW input and output

RUN No. 1 FOR UREA		INPUT VALUES		
APPL (#/AC) RATE	APPL. NO. (#/AC/YR)	URATE (#/AC/YR)	SOIL KOC	SOIL AEROBIC METABOLISM (DAYS)
81.000	10	810.000	8.0	1.0

GROUND-WATER SCREENING CONCENTRATIONS IN PPB

1.376344

A=	.167	B=	13.000	C=	-.778	D=	1.114	RILP=
-.867								
F=	-2.770	G=	.002	URATE=	810.000	GWSC=		

1.376344